

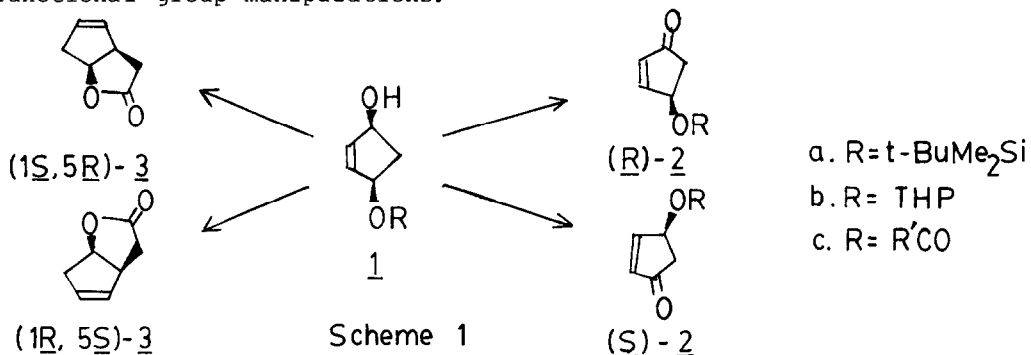
AN ASYMMETRIC SYNTHESIS OF CIS-4-t-BUTYLDIMETHYLSILOXY-2-CYCLO-
PENTEN-1-OL AND CIS-4-TETRAHYDOPYRANYLOXY-2-CYCLOPENTEN-1-OL,
VERSATILE CHIRAL SYNTHETIC INTERMEDIATE FOR PROSTANOIDS

Masatoshi Asami

Department of Chemistry, Faculty of Education,
Yokohama National University,
Tokiwadai, Hodogaya-ku, Yokohama 240, Japan

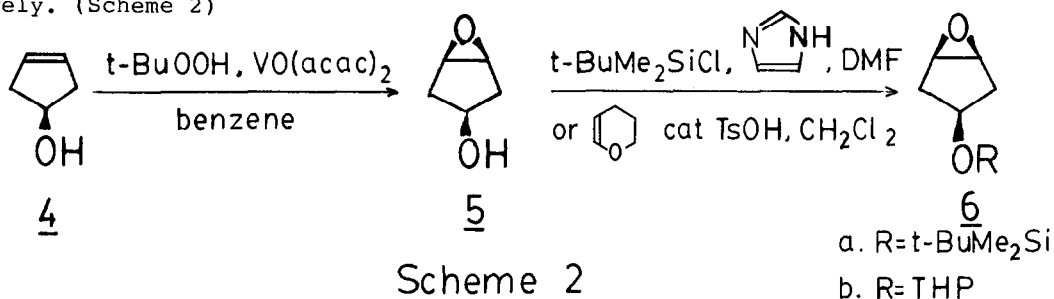
Abstract: cis-4-t-Butyldimethylsiloxy-2-cyclopenten-1-ol and cis-4-tetra-
hydropyranyloxy-2-cyclopenten-1-ol were obtained with high
enantiomeric excesses (ee) by the reaction of cis-3,4-epoxy-
cyclopentan-1-ol derivatives with chiral lithium amide. An appli-
cation to the syntheses of both (S)- and (R)-4-hydroxy-2-cyclo-
pentenone was demonstrated.

Chiral cis-4-t-butyldimethylsiloxy-2-cyclopenten-1-ol (1a) (R= t-BuMe₂Si) and cis-4-tetrahydropyranyloxy-2-cyclopenten-1-ol (1b) (R=THP) are attractive synthetic intermediates because they can be easily converted to both enantiomers of 4-hydroxy-2-cyclopentenone derivative (2) and cis-2-oxabicyclo[3.3.0]oct-6-en-3-one (3),^{1a,c,e} versatile synthetic blocks for the enantioselective synthesis of prostaglandines and various cyclopentanoid natural products.²⁾ In several reports, enantioselective enzymatic hydrolysis of cis-1,4-diacyloxy-2-cyclopentene^{1a,b,c,d} or enantioselective acylation of cis-2-cyclopenten-1,4-diol^{1e,f} was used for the synthesis of chiral cis-4-acyloxy-2-cyclopenten-1-ol (1c) (R=R'CO). 1b was derived from 1c by functional group manipulations.^{1b,c,e}



Here we wish to report the direct asymmetric synthesis of 1a and 1b from 3-cyclopenten-1-ol (4)³⁾ by applying the asymmetric reaction developed by us.⁴⁾

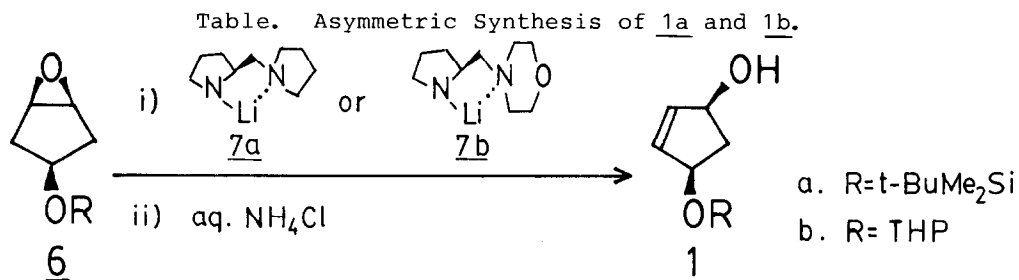
Initially, 4 was converted stereoselectively to cis-3,4-epoxycyclopentanol (5)⁵⁾ by the hydroxy-directed epoxidation with VO(acac)₂ and t-butyl hydroperoxide in benzene.⁶⁾ Then, cis-4-t-butyldimethylsiloxy-1,2-epoxycyclopentane (6a)⁷⁾ and cis-4-tetrahydropyranyloxy-1,2-epoxycyclopentane (6b)⁷⁾ were obtained by treating 5 with t-butyldimethylsilyl chloride in DMF⁸⁾ or dihydropyrane in CH₂Cl₂ in the presence of catalytic amount of p-toluenesulfonic acid⁹⁾ in 65 % or 49 % yield from 4, respectively. (Scheme 2)



Then, an asymmetric transformation of 6a to 1a was examined in various kinds of solvent with lithium (S)-2-(pyrrolidinomethyl)pyrrolidide (7a) or lithium (S)-2-(morpholinomethyl)pyrrolidide (7b). The best result was obtained in case that the reaction was carried out in benzene using 7a to yield (1S,4R)-1a⁷⁾ in 92 % with 90 % ee. 1b was also obtained with high ee. The results are summarized in Table.

A typical experimental procedure is as follows; to a benzene solution (3 ml) of (S)-2-(pyrrolidinomethyl)pyrrolidine (132 mg, 0.86 mmol) was added a hexane solution (0.5 ml) of butyllithium (0.77 mmol) under a nitrogen atmosphere at 4 °C. After 0.5 h at that temperature, 6a (102 mg, 0.48 mmol) in benzene (2 ml) was added at 4 °C and stirring was continued for 3 h. Then, aq. NH₄Cl and ether were added, and the organic layer was washed with water and brine. After drying (anhydrous Na₂SO₄) and evaporation of the solvent in vacuo, the oily substance was purified by silica-gel column-chromatography (hexane:ether=1:1) to give 1a (94 mg, 92 %), [α]_D²² +21.5 ° (c 0.94, CHCl₃).

Further, the syntheses of both (R)- and (S)-4-hydroxy-2-cyclopentenone (8) were achieved by simple functional group manipulations using (1S,4R)-1a. Oxidation of (1S,4R)-1a with pyridinium chlorochromate (PCC)¹⁰⁾ led to (R)-2a (86 %, [α]_D²⁵ +58.1 ° (c 1.13, CH₃OH); lit. [α]_D²⁰ +32 ° (c 0.051, CH₃OH) for 56 % ee of (R)-2a,^{11a)} [α]_D +62.2 °^{11b)}), which was deprotected with AcOH-THF-H₂O (3:1:1)⁸⁾ to yield (R)-8 (83 %, [α]_D²⁵ +81.3 ° (c 1.55, CHCl₃),



epoxide	lithium amide	solvent	yield/ % ^{a)}	$[\alpha]_D$ (c, CHCl ₃)	ee/ % ^{b)}
<u>6a</u>	<u>7a</u>	THF	76	$[\alpha]_D^{20} +16.5^\circ$ (1.17)	66
		ether	83	$[\alpha]_D^{22} +16.9^\circ$ (0.88)	70
		benzene	92	$[\alpha]_D^{22} +21.5^\circ$ (0.94)	90 (86) ^{c)}
		toluene	84	$[\alpha]_D^{22} +20.0^\circ$ (0.89)	84
		hexane	91	$[\alpha]_D^{22} +21.0^\circ$ (0.82)	88
<u>6a</u>	<u>7b</u>	THF	65	$[\alpha]_D^{22} +5.7^\circ$ (0.69)	26
		benzene	78	$[\alpha]_D^{16} +19.7^\circ$ (0.68)	89
<u>6b</u>	<u>7a</u>	THF	89	$[\alpha]_D^{19} +23.2^\circ$ (0.82)	62 ^{d)}
		benzene	77	$[\alpha]_D^{18} +27.3^\circ$ (0.77)	89 ^{d)}

a) Isolated yield.

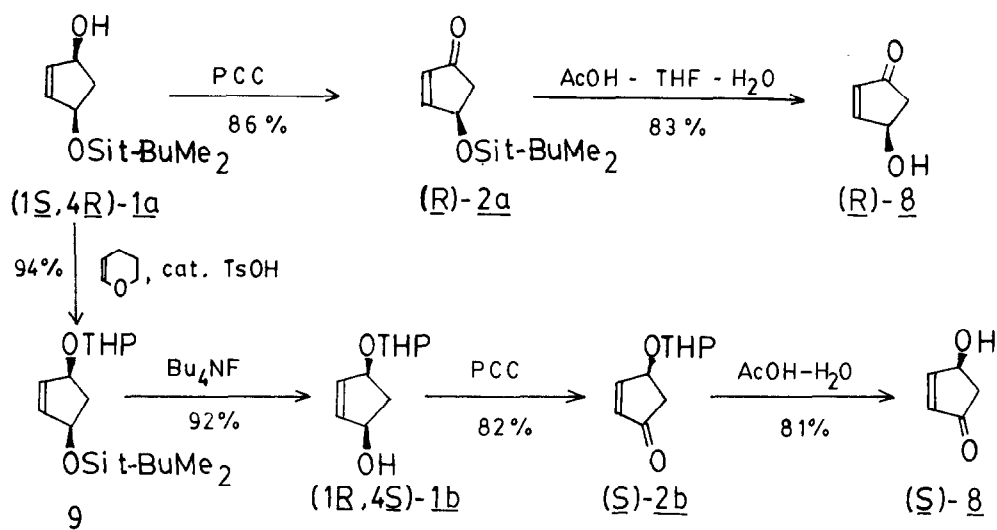
b) Determined by conversion to (4R,1S)-4-hydroxy-2-cyclopentenyl benzoate.^{1e)}

c) Determined by conversion to (S)- and (R)-4-acetoxy-2-cyclopentenone followed by ¹H NMR analysis in the presence of Eu(hfbc)₃.

d) Optical rotation value of 1b has been reported ($[\alpha]_D^{20} -20.3^\circ$ (c 1.3, CHCl₃),^{1e)} $[\alpha]_D^{20} +21.5^\circ$ (c 3.12, CHCl₃) for 86 % ee of (1S,4R)-1b^{1c)}, but it would not be possible to calculate the ee based on these value because there is another chiral center in the tetrahydropyrane ring in 1b.

$[\alpha]_D^{26} +83.5^\circ$ (c 2.00, CH₃OH); lit. $[\alpha]_D^{20} -94.1^\circ$ (c 3.4, CHCl₃) for (S)-8,^{1e)} $[\alpha]_D^{24} +83.1^\circ$ (c 1.70, CH₃OH) for 94 % ee of (R)-8¹²⁾. (S)-8 ($[\alpha]_D^{24} -79.3^\circ$ (c 1.5, CHCl₃), $[\alpha]_D^{28} -85.1^\circ$ (c 1.34, CH₃OH)) was obtained as follows; i) conversion of (1S,4R)-1a into (1R,4S)-1-t-butyldimethylsiloxy-4-tetrahydropyranyloxy-2-cyclopentenol (9)⁷⁾ (dihydropyrane, cat. p-toluene-sulfonic acid, CH₂Cl₂;⁹⁾ 94 %, $[\alpha]_D^{24} +2.04^\circ$ (c 1.08, CHCl₃)), ii) removal of t-butyldimethylsilyl group (Bu₄NF, THF)⁸⁾ leading to (1R,4S)-1b (92 %, $[\alpha]_D^{27} -22.3^\circ$ (c 1.25, CHCl₃); lit. $[\alpha]_D^{20} -20.3^\circ$ (c 1.3, CHCl₃),^{1e)} $[\alpha]_D^{20} +21.5^\circ$ (c 3.12, CHCl₃) for 86 % ee of (1S,4R)-1b^{1c)}); iii) oxidation (PCC) of (1R,4S)-1b to (S)-2b (82 %, $[\alpha]_D^{29} -54.9^\circ$ (c 0.97, CHCl₃); lit.^{1e)} $[\alpha]_D^{20} -70.5^\circ$ (c 1.3, CHCl₃)); iv) followed by deprotection (AcOH-H₂O, (7:3); 81 %).^{1e)} (Scheme 3)

Both (R)- and (S)-8 were derived to (R)- and (S)-4-acetoxy-2-cyclo-



pentenone, and the ee of those were 86 % by ^1H NMR spectra taken with $\text{Eu}(\text{hfbc})_3$.¹²⁾

It should be noted that the chiral cyclopentenol derivative 1, a useful synthetic intermediate for the enantioselective synthesis of various cyclopentanoid natural products, was easily obtained with high ee by the reaction of achiral epoxide with chiral lithium amide.

References

- 1) a) S. Takano, K. Tanigawa, and K. Ogasawara, *J. Chem. Soc. Chem. Commun.*, 1976, 189.
 b) Y.-F. Wong, C. S. Chen, G. Girdaukas, and C. J. Sih, *J. Am. Chem. Soc.*, 106, 3695, (1984). c) K. Laumen and M. Schneider, *Tetrahedron Lett.*, 25, 5875 (1984). d) K. Laumen, E. H. Reimerdes, M. Schneider, and H. Görisch, *ibid.*, 26, 407 (1985). e) M. Nara, S. Terashima, and S. Yamada, *Tetrahedron*, 36, 3161 (1980). f) L. Duhamel and T. Herman, *Tetrahedron Lett.*, 26, 3099 (1985).
- 2) a) M. Harre, P. Raddatz, R. Walenta, and E. Winterfeldt, *Angew. Chem.*, 94, 496 (1982).
 b) R. Noyori and M. Suzuki, *ibid.*, 96, 854 (1984). c) J. J. Partridge, N. K. Chada, and M. R. Uskoković, *J. Am. Chem. Soc.*, 95, 7171 (1973). d) L. Gruber, I. Tömösközi, E. Major, and G. Kovacs, *Tetrahedron Lett.*, 1974, 3729. e) I. Tömösközi, L. Gruber, G. Kovacs, I. Szekely, and V. Simonidesz, *ibid.*, 1976, 4639. f) H. Nagaoka, T. Miyakoshi, and Y. Yamada, *ibid.*, 25, 3621 (1984).
- 3) H. M. Hess and H. C. Brown, *J. Org. Chem.*, 32, 4138 (1967).
- 4) M. Asami, *Chem. Lett.*, 1984, 829.
- 5) a) A. C. Darby, H. B. Henbest, and I. McClenaghan, *Chem. Ind. (London)*, 1962, 462.
 b) F. David, *J. Org. Chem.*, 46, 3512 (1981).
- 6) a) K. B. Shapless, R. C. Michaelson, *J. Am. Chem. Soc.*, 96, 5254 (1974). b) G. Stork, I. Paterson, and F. K. C. Lee, *ibid.*, 104, 4686 (1982).
- 7) All new compounds showed satisfactory IR and ^1H NMR spectra.
- 8) E. J. Corey and A. Venkateswarlu, *J. Am. Chem. Soc.*, 94, 6190 (1972).
- 9) K. F. Bernady, M. B. Floyd, J. F. Poletto, M. J. Weiss, *J. Org. Chem.*, 44, 1438 (1979).
- 10) E. J. Corey and J. W. Suggs, *Tetrahedron Lett.*, 1975, 2647.
- 11) a) T. Tanaka, S. Kurozumi, S. Miura, M. Koboyashi, and S. Ishimoto, *Tetrahedron*, 32, 1713, 1893 (1976). b) J. Nokami, T. Ono, S. Wakabayashi, A. Hazato, and S. Kurozumi, *Tetrahedron Lett.*, 26, 1985 (1985).
- 12) R. Noyori, I. Tamino, M. Yamada, and M. Nishizawa, *J. Am. Chem. Soc.*, 106, 6717 (1984).

(Received in Japan 1 August 1985)